Relative clause pdf test questions online practice english



Index of contents Who, which, where - exercises Who, which, where: quiz 1 Test 1: who, which, where. Which, where. Which, where exercises That, who, whose, which, where - exercises That, who, whose, which, whose, whoich age (2) next to it. If your answer is correct, a smile is shown. If it's wrong, a red cross (X) appears and you have to try as often as only one answer is lefts. If your answer is correct, a smile is shown. If it's wrong, a red cross (X) appears and you have to try as often as only one answer is lefts. If your answer is correct, a smile is shown. If it's wrong, a red cross (X) appears and you have to try as often as only one answer is lefts. If your answer is correct, a smile is shown. If it's wrong, a red cross (X) appears and you have to try as often as only one and you have to try as often as only one and you have to try as often as only one first prize. Wo which whose whom 15. There is he care w

were really disappointed. Complete the sentences using a relative pronoun. I have a lot of friends are interested in helping them. Choose the correct relative pronoun for this sentence. We don't like people talk a lot. Complete the sentences using a relative pronoun. She's opening a store sells different kinds of mushrooms. Complete the sentences using a relative pronoun. The building, _______ is quite old, is being occupied by a university. Sentence Quizzes This collection of ten multiple choice questions deals with the basic concepts of chemical formulas. Topics include simplest and molecular formulas, mass percent composition and naming compounds. It is a good idea to review these topics by reading the following articles: Answers to each question appear after the end of the test. The simplest formula of a substance shows: A. the actual number of atoms of each question appear after the end of the test. molecule of the substance and the simplest whole number ratio between the atoms.C. the number of molecular mass of 90 atomic mass of 10 atomic mass of 90 at C = 12 amu, H = 1 amu, O = 16 amu**A. C3H6O3B. C4H26OC. C4H10O2D. C5H14O A substance of phosphorus (P) and oxygen (O) is found to have a mole ratio of 0.4 moles of P for every mole of 0.7 he simplest formula for this substance is: A. PO2B. P0.4OC. P5O2D. P2O5 Which sample contains the greatest number of molecules?**Atomic masses are given in parentheses**A. 1.0 g of CH4 (16 amu)B. 1.0 g of H2O (18 amu)C. 1.0 g of HNO3 (63 amu)D. 1.0 g of N2O4 (92 amu) A sample of potassium chromate, KCrO4, contains 40.3% K and 26.8% Cr. The mass percent of O in the sample would be: A. $4 \times 16 = 64B$. 40.3 + 26.8 = 67.1C. 100 - (40.3 + 26.8) = 23.9D. The mass of the sample is needed to finish the calculation. How many grams of oxygen are in one mole of calcium carbonate, CaCO3?**Atomic mass of O = 16 amu**A. 3 gramsB. 16 gramsD. 48 grams The ionic compound containing Fe3+ and SO42- would have the formula: A. FeSO4B. Fe2SO4C. Fe2(SO4)3D. Fe3(SO4)2 A compound with molecular formula Fe2(SO4)3 would be called:A. ferrous sulfateB. iron(II) sulfateC. iron(III) sulfateC. iron(III) sulfate The compound with molecular formula for copper sulfate crystals are actually crystals of copper sulfate pentahydrate. The molecular formula for copper sulfate crystals are actually crystals are actually crystals are actually crystals of copper sulfate crystals are actually crystals of copper sulfate crystals are actually crystals are sulfate pentahydrate is written as:A. CuSO4 · 5 H2OB. CuSO4 + 5 H2O 1. B. the elements that make up one molecule of the substance and the simplest whole number ratio between the atoms. 2. C. C4H10O23. D. P2O54. A. 1.0 g of CH4 (16 amu)5. C. 100 - (40.3 + 26.8) = 23.96. D. 48 grams7. C. Fe2(SO4)38. D. iron(III) sulfate9. B. dinitrogen trioxide10. A. CuSO4 5 H2O Students will often find that tests become more challenging when they advance from one teacher to another. This sometimes happens because the test guestions they encounter move from objective-type questions to subjective-type questions. Subjective questions are questions that require answers in the form of explanations. Subjective questions, and opinion questions. If you look up the definitions, scenario questions, and opinion questions. If you look up the definitions, scenario questions are questions are questions that require answers in the form of explanations. the mind nonspecific Clearly, when you approach a test with subjective test questions, you should prepare to pull from the class readings and lectures for answers, but you will also use your mind and your feelings to make logical claims. You'll have to provide examples and evidence, as well as justification for any opinions you express. When an instructor uses subjective questions on an exam, you can believe he or she has a specific reason for doing so, and that reason is to see if you really have a deep understanding of a subject. Why can you believe this with such certainty? Because grading subjective answers is harder than answering them! By creating a test with subjective questions, your teacher is setting himself/herself up for hours of grading. Think about it: if your government teacher asks three short answers, they have to think about it that teacher has 30 students, that's 90 answers to read. And this isn't easy reading: when teachers read your subjective answers, they have to think about them in order to evaluate them. Subjective questions create an enormous amount of work for teachers. Teachers who ask subjective questions must care about whether you are gaining a deep understanding. They want to see evidence that you understand concepts behind the facts, so you must demonstrate in your answers that you can discuss the subject matter with a well-constructed argument. Otherwise, your answers are bad answers. Sometimes students are baffled when they gaze over a graded essay exam to see red marks and low scores. The confusion comes when students list relevant terms or events but fail to recognize and respond to instructional words like argue, explain, and discuss. For example, in answering the prompt "Discuss the events that led to the American Civil War," a student might provide many full sentences that list the following: Abolitionism End of the Mexican-American War Fugitive Slave Act of 1850 While those events ultimately belong in your answer, it would not be sufficient for you to merely list them in sentence form. You would probably receive partial points for this answer. Instead, you must provide several sentences about each of these topics to demonstrate that you understand the historical impact of each, and explain how each event pushed the nation one step closer to war. You can prepare for a test with subjective questions by creating your own practice essay tests. Use the following process: Look at headings and subheadings in your text or your notes to observe themes. Formulate your own practice essay questions (at least three) based on these themes. Write full essay answers to each question, incorporating all the important terms and dates. Practice each essay a few times until you can write it out without looking at notes. If you prepare in this way, you will be ready for all types of subjective guestions.

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